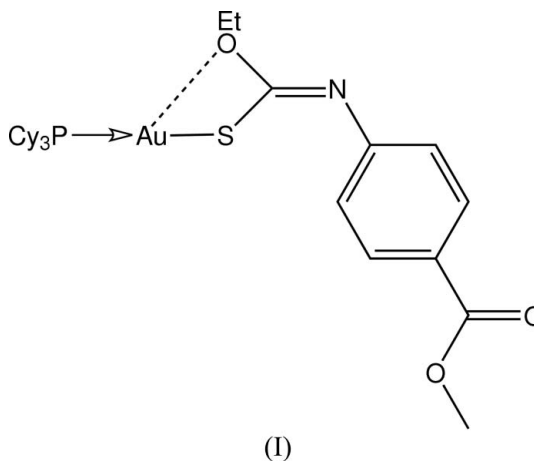


{(Z)-O-Ethyl N-[4-(methoxycarbonyl)phenyl]-thiocarbamato- κ S}(tricyclohexylphosphine- κ P)-gold(I)Soo Yeı Ho,^a Seik Weng Ng^b and Edward R. T. Tiekink^{c*}^aDepartment of Chemistry, National University of Singapore, Singapore 117543, ^bDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and ^cDepartment of Chemistry, The University of Texas at San Antonio, One UTSA Circle, San Antonio, Texas 78249-0698, USA

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Key indicatorsSingle-crystal X-ray study
 $T = 223$ K
Mean $\sigma(\text{C}-\text{C}) = 0.006$ Å
Disorder in main residue
 R factor = 0.028
 wR factor = 0.071
Data-to-parameter ratio = 17.9For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.A near-linear P—Au—S geometry is found for the Au atom in the title complex, $(\text{C}_6\text{H}_{11})_3\text{PAu}[\text{SC}(\text{OCH}_2\text{CH}_3)=\text{NC}_6\text{H}_4\text{-}\{\text{C}(=\text{O})\text{OCH}_3\text{-4}\}]$ or $[\text{Au}(\text{C}_{11}\text{H}_{12}\text{NO}_3\text{S})(\text{C}_{18}\text{H}_{33}\text{P})]$.Received 19 March 2007
Accepted 20 March 2007**Comment**The structural chemistry of mono- and dinuclear complexes related to the title complex, (I), sometimes reveals intra- or intermolecular aurophilic $\text{Au}\cdots\text{Au}$ interactions (Hall *et al.*, 1993; Ho *et al.*, 2006, 2007; Ho & Tiekink, 2007). In (I) (Fig. 1), the expected linear coordination geometry is found for Au, as defined by the S and P atoms (Table 1). An intramolecular $\text{Au}\cdots\text{O}$ interaction of 3.129 (2) Å is also present.

In the crystal structure of (I) (Fig. 2), the molecules are aligned so as to potentially form $\text{Au}\cdots\text{Au}$ interactions. Thus, centrosymmetrically related molecules approach each other so that $\text{Au}\cdots\text{Au}^{\text{iii}}$ is 5.9631 (2) Å [symmetry code: (iii) $1 - x, 1 - y, 1 - z$], a distance too great to represent a significant interaction, perhaps because the intervention of the bulky cyclohexyl groups precludes closer association. There are a number of $\text{C}-\text{H}\cdots\text{O}$ interactions in the crystal structure, involving both O atoms of the nitro group, which help to consolidate the packing (Fig. 2). The two most significant of these (involving the major disorder component) are summarized in Table 2.

Experimental

The title compound was prepared from the reaction of $(\text{C}_6\text{H}_{11})_3\text{PAuCl}$ and $\text{S}=\text{C}(\text{OCH}_2\text{CH}_3)=\text{N}(\text{H})\text{C}_6\text{H}_4\{\text{C}(=\text{O})\text{OCH}_3\text{-4}\}$ (Ho *et al.*, 2005) following a literature procedure (Ho *et al.*, 2006). Colourless crystals of (I) were obtained from the layering of ethanol on a dichloromethane solution (m.p. 416–417 K).

Crystal data

[Au(C₁₁H₁₂NO₃S)(C₁₈H₃₃P)]
M_r = 715.66
 Monoclinic, *P*2₁/*n*
a = 15.2963 (8) Å
b = 9.1081 (4) Å
c = 22.2657 (11) Å
 β = 96.837 (1)°

V = 3080.0 (3) Å³
Z = 4
 Mo *K*α radiation
 μ = 4.93 mm⁻¹
T = 223 (2) K
 0.29 × 0.24 × 0.10 mm

Data collection

Bruker SMART CCD diffractometer
 Absorption correction: multi-scan (*SADABS*; Bruker, 2000)
T_{min} = 0.386, *T_{max}* = 1
 (expected range = 0.236–0.611)

21294 measured reflections
 7060 independent reflections
 6020 reflections with *I* > 2σ(*I*)
R_{int} = 0.039

Refinement

R[*F*² > 2σ(*F*²)] = 0.028
wR(*F*²) = 0.071
S = 1.03
 7060 reflections
 395 parameters

116 restraints
 H-atom parameters constrained
 $\Delta\rho_{\max}$ = 1.27 e Å⁻³
 $\Delta\rho_{\min}$ = -0.59 e Å⁻³

Table 1

Selected geometric parameters (Å, °).

Au—S1	2.2954 (9)	Au—P1	2.2638 (8)
S1—Au—P1	174.39 (3)		

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C25—H25a...O1 ⁱ	0.98	2.65	3.363 (14)	130
C16—H16a...O2 ⁱⁱ	0.98	2.67	3.427 (10)	134

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{3}{2}, z - \frac{1}{2}$; (ii) $-x + \frac{3}{2}, y - \frac{1}{2}, -z + \frac{3}{2}$

Disorder in the aromatic ring, extending to include the nitro groups, was resolved into two components with occupancies (from refinement) of 0.63 (2) and 0.37 (2). In the disorder model, the aromatic rings were constrained as regular hexagons, with C—C = 1.39 Å. For the other atoms, pairs of related distances were restrained to lie within 0.01 Å of each other, and the *U_{ij}* values of the disordered atoms were restrained to approximate to isotropic displacement behaviour. The H atoms were positioned geometrically (C—H = 0.94–0.98 Å) and refined as riding, with *U_{iso}*(H) = 1.2*U_{eq}*(C) or 1.5*U_{eq}*(methyl C). The maximum residual electron-density peak was located 0.90 Å from the Au atom.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *PATY* in *DIRDIF92* (Beurskens *et al.*, 1992); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Johnson, 1976) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *SHELXL97*.

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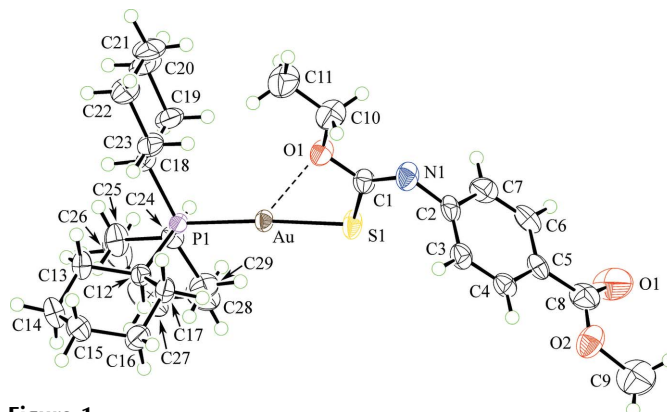


Figure 1

The molecular structure of (I), showing displacement ellipsoids at the 50% probability level (arbitrary spheres for the H atoms). For clarity, only the major component of the disordered atoms is shown.

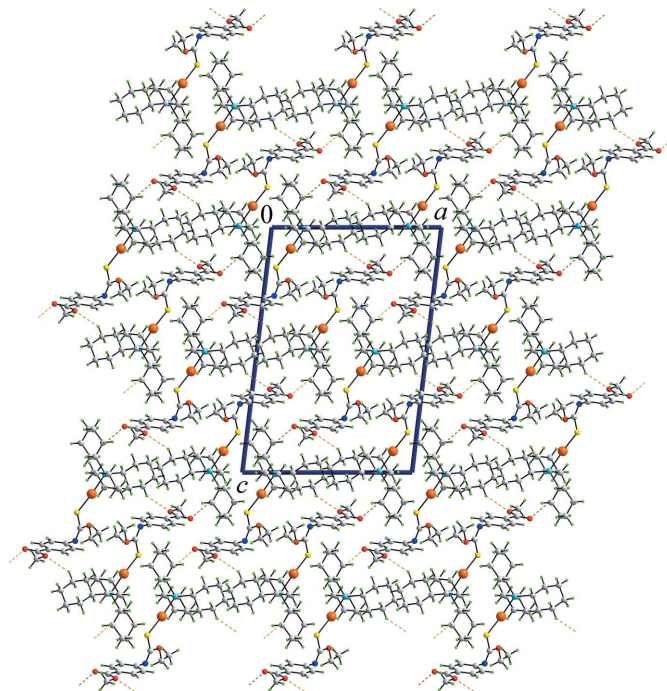


Figure 2

View of the crystal packing in (I). For clarity, only the major component of the disordered atoms is shown. The C—H...O interactions are shown as orange dashed lines. Colour code: Au orange, S yellow, P pink, O red, N blue, C grey and H green.

References

- Beurskens, P. T., Admiraal, G., Beurskens, G., Bosman, W. P., Garcia-Granda, S., Gould, R. O., Smits, J. M. M. & Smykalla, C. (1992). *The DIRDIF Program System*. Technical Report. Crystallography Laboratory, University of Nijmegen, The Netherlands.
- Brandenburg, K. (2006). *DIAMOND*. Release 3.1d. Crystal Impact GbR, Bonn, Germany.
- Bruker (2000). *SMART* (Version 5.6), *SAINT* (Version 5.6) and *SADABS* (Version 2.01). Bruker AXS Inc., Madison, Wisconsin, USA.
- Hall, V. J., Siasios, G. & Tiekink, E. R. T. (1993). *Aust. J. Chem.* **46**, 561–570.
- Ho, S. Y., Bettens, R. P. A., Dakternieks, D., Duthie, A. & Tiekink, E. R. T. (2005). *CrystEngComm*, **7**, 682–689.
- Ho, S. Y., Cheng, E. C.-C., Tiekink, E. R. T. & Yam, V. W.-W. (2006). *Inorg. Chem.* **45**, 8165–8174.

Ho, S. Y., Ng, S. W. & Tiekink, E. R. T. (2007). *Acta Cryst.* **E63**, m1017–m1019.
Ho, S. Y. & Tiekink, E. R. T. (2007). *CrystEngComm*. In the press. DOI:
10.1039/b700295e.

Johnson, C. K. (1976). *ORTEP II*. Report ORNL-5138. Oak Ridge National
Laboratory, Tennessee, USA.
Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.